

GUJARAT TECHNOLOGICAL UNIVERSITY
B.PHARM - SEMESTER- 3 EXAMINATION – WINTER -2019

Subject Code: 2230004**Date: 25-11-2019****Subject Name: Pharmaceutical Chemistry-IV (Organic Chemistry – I)****Time: 10:30AM TO 01:30PM****Total Marks: 80****Instructions:**

1. Attempt any five questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

- | | | |
|-------------|---|-----------|
| Q.1 | (a) Explain Different types of bond with example of each one. | 06 |
| | (b) Define Carbocation. Write down the generation and reactions of Carbocation. | 05 |
| | (c) Write a note on electronegativity and Polarity. | 05 |
| Q.2 | (a) Discuss in detail how Grignard reagent is helpful for synthesis of alcohol and alkanes. | 06 |
| | (b) Write a descriptive note on Molecular Orbital Theory. | 05 |
| | (c) How will you distinguish between primary, secondary and tertiary alcohol. | 05 |
| Q.3 | (a) Explain the Principle of Dumas Method and Kjeldahl method. | 06 |
| | (b) Define Hybridization. Explain sp^2 hybridization in detail. | 05 |
| | (c) Discuss the chemical preparation and reactions of ether. | 05 |
| Q.4 | (a) Discuss the Haworth Synthesis of Anthracene. | 06 |
| | (b) Explain halogenation of methane with its mechanism. | 05 |
| | (c) Give two methods of preparation of alkene and alkyl halide. | 05 |
| Q.5 | (a) Give detailed notes on Aldol Condensation reaction with its mechanism. | 06 |
| | (b) Write short note on Resonance. | 05 |
| | (c) Write down in detail on Friedal Craft alkylation reaction in detail. | 05 |
| Q. 6 | (a) Write a note on determination of molecular weight of volatile substances. | 06 |
| | (b) Discuss quantitative analysis of Nitrogen element. | 05 |
| | (c) State Saytzeff's orientation and Kharash Peroxide effect with examples. | 05 |
| Q.7 | (a) Write down the methods of preparation and reactions of Diene. | 06 |
| | (b) Write a note on Bonding and Anti-bonding orbitals. | 05 |
| | (c) Explain the Intermolecular and Intramolecular forces. | 05 |
